# LETTERS TO THE EDITOR

#### TO THE EDITOR

A recent R & D Note by Luus and Jaakola (1973) references a plant that has been a choice example for study by different optimal research schemes. The primary reference for defining the degrees of freedom for the material balance equations and their ordering and substitution of variables suggested by Luus (1937b) is Christiansen (1970) who disposed of this question for a general class of equations of interest in optimal plant design. The fact that the optimal location was unaffected by changes in FP for the steady state plant was indicated by Williams and Otto (1960) and reiterated by Mason (1971) and Mason and Crosser (1972).

We list all references to papers shown in Figure 2. using this plant which we know of so that other workers may use results 1.19 remind us that the chemical reac-previously obtained and published tions of this plant are irreversible and Figure 1 shows the ROIS for constant feasible solutions must be at larger FRC and T for the investment used values of y and  $\phi y$ . This surface shows by Adleman (1972), Alghren (1966), a max at  $\phi = 0$  corresponding to total Dibella (1965), Luus (1973), Ray recycle of all byproduct (practical = 4763. Important singularities are expense might permit this). the steep cliff near FG = 0 ( $FG \ge 0$ up to the right), the singularity lo- of investment (I) correlation, such as cated by the broken line, and the that of Dibella and Stevens (1965) or steep negative lobe located between Mason (1971). Although the location them. Search procedures can be confused by the steep ridge or obtain the undesirable negative lobe which is positive for other choices of equations for the investment (Mason, 1971). At lower values of T and FRC the ridge maximum shifts along the cliff to the 0.4left, but the surface shape remains as shown. We especially feel that the plots (Figure 1) of the response surface will be of value to those who wish to understand the relative success of optimal search schemes for this plant and to realize that the location of the optimum is sensitive to the correlation used for the investment.

Because the equations can be ordered into a straight-forward (direct) calculational scheme, the equation for the ROI can be obtained in terms of the set of 4 (or 5 if FP is included) independent variables. While the ex- 2.0 pression is lengthy, and a reduced form depends upon the correlation chosen for the investment, we have # studied two cases in some analytical detail.

In case I studies, the investment is assumed to be constant, which corresponds to the case of an operating Fig. 2. Contours of revenue as function of mplant already constructed. This turns

out to be interesting in addition be- of the optimum is altered by the I cause CHESS-UMR simulations show correlation, the analysis is unaffected, the investment changes less than 10% and only the details for investment over the range of positive return. The proportional to FR (Mason (1972)) resulting equation, which shows the will be presented. For this case at connet profit, for constant FRC and T is

stant T and FRC and for  $\phi = .4$  the

$$\frac{10^{-8}R = 2.71\phi^{3}y^{3} + (-.737\phi^{3} + 9.35\phi^{2})y^{2} + (.850\phi^{2} - 7.35\phi - .034\phi^{3})y - .253\phi + .04\phi^{2}}{y(y\phi - 1.19)}$$
(1)

where  $y = 10^{-4}$  FRE,  $\phi = FD/FS$  equation is (unity minus the flow recycle ratio). This is shown for R > 0 in Figure 2, where the presentation is simplified by replacing the original coordinates of Figure 1 with  $m = \phi y$  as a function of  $\phi$ . The curved ridge of Figure 1 this ratio of cubics can be resolved to then becomes essentially horizontal as

The singularities at y = 0,  $y\phi =$ (1971), and Vinturella (1968) for FP considerations suggest that operating

Case II studies include some kind

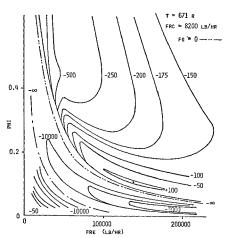
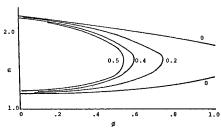


Fig. 1. Contours of return on investment-Adleman (1972).



and o.

$$\frac{-y^3 + 3.340y^2 - 2.580y - 0.0355}{y^3 - 1.140y^2 - 0.041y + 0.0965}$$
$$= 2.24 Z \quad (2)$$

$$\frac{0.559}{y + 0.270} + \frac{0.480}{y - 0.320} - \frac{0.0425}{y - 1.097} = Z \quad (3)$$

where  $Z = 10^{-4} ROI$ ,  $y = 10^{-4} FRE$ , PHI = 0.4, FRC = 1500 lb/hr, and T = 580 R which illustrates the singularities and adjacent max pictured in Figure 1.

The behavior of the surface in FRC and T is contained in the coefficients of Equation (2). Straight-forward analysis, for Equation (1), of the most important terms of the polynomial shows that the extreme must lie at the lowest T allowed and for the lowest value of FRC. Although the maximum is unattractive physically, it is an excellent point from which to begin the optimal search of a more complete representation such as that of CHESS.

Approximations, simulations, analysis of this kind are valuable when they reduce the total time and expense of calculation. Indeed, when Whitehead (1973) maintains  $FB \ge 0$  in his calculations for the material balance, he defines an even narrower region on this surface. The dominant contribution to the ridge is the numerator of Equation (2) indicating that it is the net profit, and not the investment, which is the controlling feature of this plant.

#### LITERATURE CITED

Adelman, A., and W. F. Stevens, "Process Optimization by the 'Complex' Method," AIChE J., 18, 20 (1972). Ahlgren, T. D., and W. F. Stevens, "Process Optimization in the Presence of Error," Ind. Eng. Chem. Process Design Development, 5, 290 (1966).

Optimization," AIChE J., 16, 177 (1970).

Dibella, C. W., and W. F. Stevens, "Process Optimization by Nonlinear Programming," Ind. Eng. Chem. Process Design Development, 4, 16 (1965).

Luus, R., and T. H. I. Jaakola, "Optimization by Direct Search and Systematic Reduction of the Size of Search Regions," AIChE J., 19, 760 (1973a).

ibid., 645.

Mason, J. T., III., "Response Surface Study of a Characteristic Chemical Plant, Ph.D. dissertation, Univ. of Missouri— Rolla (1971).

Surface of a Small Chemical Plant," paper presented at 71st National Meeting of Am. Inst. Chem. Engrs., Dallas, Texas (1972).

Ray, W. H., B. S. Jung, and W. Mirosh, "Large Scale Process Optimization Techniques Applied to Chemical and Petro-leum Processes," Can. J. Chem. Eng.,

49, 844 (1971). Vinturella, J. B., "Chemical Process Design and Optimization by Non-Linear Programming," Ph.D. dissertation, Tulane University, Baton Rouge (1968).

Whitehead, B. F., Ph.D. thesis, University of Leeds, England (1973).

Williams, T. J., and R. E. Otto, "A Generalized Chemical Processing Model for the Investigation of Computer Control," Am. Inst. Elec. Engr. Trans., 791, 458

(1960).

CHEMICAL ENGINEERING University of Missouri-Rolla Rolla, Missouri 65401

## TO THE EDITOR

In a recent Letter to the Editor, Dyer (1973) has as a "main objective . . . to set straight the common error made in using Fick's law for sublimation dehydration (freeze-drying) problems." He maintains that this no relationship to those determined error is failure to account for the bulk from direct experimental measureflow due to a gradient in total pressure ments. and indicates that this omission has

scured the reasoning behind the mass-tributing factor. The definition of  $D_{te}$  transfer equation used by Sandall et al. adopted by Dyer (1973) is proper (1969) allows fully for bulk flow due to contributes. Further,  $D_{te}$  cannot readily a gradient in total pressure, as well be related to fundamentally different expression is given by Mason et al. in the analysis of Gunn and King.

Christensen, J. H., "Structure of Process (1967) in a very different form. Both This essential difference is not related Mason et al. and Gunn and King to inherent differences between the demonstrate experimental verification dusty-gas and capillary models of of the analysis over a wide range of porous media. Wakao et al. (1965) conditions covering gradients in both obtain a prediction of simultaneous total pressure and composition.

this general equation could be greatly of Gunn and King and Mason et al. simplified for freeze-drying of meat, for the dusty-gas model. yielding Equation (16) of Sandall et The conditions unde Optimization of a Complex System," al. (1967) for conditions where the transfer limitations become important rate-limiting effects of mass transfer in freeze-drying should also be conrate-limiting effects of mass transfer in freeze-drying should also be conbecome important. Data presented in Figure 10 of Sandall et al. bear out the form of this simplified equation. A survey of the available published Cox and Dyer (1972) but can be a ., and O. K. Crosser, "Optimal transport data for freeze-dried foods significant rate limit in the presence Search on the Return-on-Investment convinced the authors that this approximation is valid for a much wider range of freeze-dried products than just poultry meat. However, ceivably drying conditions and transport coefficients may occur for which the simplified relationship is invalid. In this case it is only necessary to utilize the general relationship referred to above at the cost of considerable increased complexity.

It has been shown (Gunn and King, 1971) that experimental freeze-drying rate curves for meat can be predicted with good accuracy using the simplified equation. In these calculations only one constant, the external mass transfer coefficient, was fitted to the experimental drying data. The relevant transport coefficients were measured J. T. Mason, III directly on the same samples of meat AND previously freeze-dried by Sandall et Cox, C. C., and D. F. Dyer, "Freeze-sser al. (1967). Effectiveness factors for Drying of Spheres and Cylinders," O. K. Crosser al. (1967). Effectiveness factors for diffusion were determined from the high pressure isobaric counter-diffusion of nitrogen-helium mixtures. Effective Knudsen diffusivities were determined from low pressure permeation measurements of pure nitrogen. This procedure is very different from that of force-fitting an equation with an adjustable, pressure-dependent transport coefficient to drying data. In the latter case an incorrect mathematical model can be fitted to the data, but the calculated transport coefficients will bear

The division between diffusional been made by Sandall et al. (1967). flow and bulk flow is somewhat arbi-Publication delays may have obt trary when Knudsen transport is a con-Their effective internal mass-transfer when bulk diffusion and viscous flow coefficient D' is based upon concurrent alone are involved, but results in D<sub>te</sub> work by Gunn and King (1969, 1971). being dependent upon both the level The general expression presented as and gradient of absolute pressure in Equation (8) by Gunn and King a complex way when Knudsen flow as other viscous, diffusive, and Knud- independent measurements of transport sen flow effects. An entirely equivalent coefficients, as can the parameters used

diffusion and flow from the capillary Gunn and King (1971) showed that model which is close to the result

> The conditions under which masssidered. Mass transfer external to the food is neglected in the analyses given by Dyer and Sunderland (1968) and of substantial partial pressures of inerts and/or when there are constrictions in vapor flow paths within the chamber. Analysis of the simultaneous effects of mass and heat transport shows that mass transfer within the dried shell of the food becomes a significant rate limit for two conditions-for substantial partial pressures of inerts, in which case the simplified equation of Sandall et al. is valid, and for very low frozenzone temperatures during drying, in which case the Knudsen diffusion term dominates with a significant contribution from viscous flow also being possible in some cases. If inerts are absent in the latter case, Equation (7) from Gunn and King (1971) is applicable.

### LITERATURE CITED

J. Heat Trans., 94, 57 (1972).

Dyer, D. F., Letter to the Editor, AIChE J., 19, 670 (1973).

-., and J. E. Sunderland, "Heat and Mass Transfer Mechanisms in Sublimation Dehydration," J. Heat Transfer, 90, 379 (1968).

Gunn, R. D., and C. J. King, "Mass Transport in Porous Materials under Com-

bined Gradients of Composition and Pressure," AIChE J., 15, 507 (1969). Sandall, O. C., C. J. King, and C. R. Wilke, "The Relationship Between Transport Properties and Rates of France Descriptions of the Properties of Properties and Rates of France Descriptions of the Properties and Rates of France Descriptions of the Properties and Rates of France Descriptions of the Properties and Rates of Properties and Rate Freeze-Drying of Poultry Meat," ibid.,

13, 428 (1967).

Mason, E. A., A. P. Malinauskas, and R. B. Evans, "Flow and Diffusion of Gases in Porous Media," J. Chem. Phys., 46, 3199 (1967).

Vakao, N., S. Ontani, and J. M. Smith, Significance of Pressure Gradients in Porous Materials," AIChE J., 11, 435, 439 (1965).

> ROBERT D. GUNN DEPT. OF CHEMICAL ENGINEERING University of Wyoming LARAMIE, WYOMING 82070 AND

C. JUDSON KING DEPT. OF CHEMICAL ENGINEERING University of California Berkeley, California 94720